

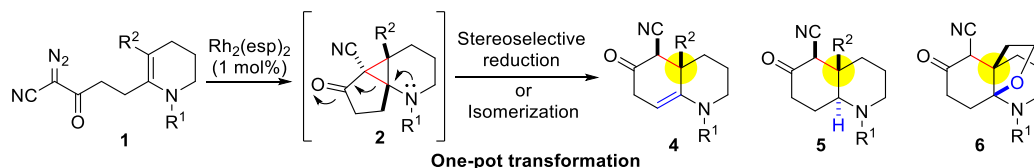
Title	Synthesis of Octahydro- and Decahydroquinolines by a One-Pot Cascade Reaction of Tetrasubstituted Enecarbamate
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# Synthesis of Octahydro- and Decahydro-Quinolines by a One-Pot Cascade Reaction of Tetrasubstituted Enecarbamate

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Supporting Information Placeholder

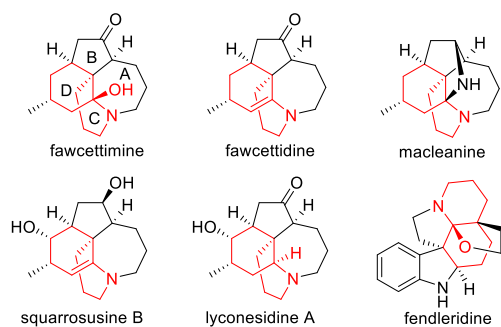


**ABSTRACT:** A transition metal catalyzed-cyclopropanation followed by ring-opening was investigated for the synthesis of octahydroquinolines **4** and decahydroquinolines **5** having a quaternary carbon center at the angular position, which are core structures of the fawcettimine-type alkaloids. A tandem reaction was also established for the synthesis of decahydroquinolines **5** and the tricyclic compound **6** through an iminium ion intermediate, readily produced by acidic treatment of cyclopropane **2**.

Fawcettimine, isolated from *Lycopodium fawcetti* by Burrell and co-workers in 1959, is one of the representative *Lycopodium* alkaloids (Figure 1).<sup>1</sup> It is one of more than 80 congeners reported to date, that include fawcettidine,<sup>1,2</sup> macleanine,<sup>3</sup> and squarrosusine B,<sup>4</sup> each of which belong to the fawcettimine class of alkaloids. Recently, new members of this alkaloid class, such as lyconesidines,<sup>5</sup> have been found in related species. Among these compounds, a complex tetracyclic skeleton is common, and the structural diversity mainly derives from the oxidation level and substituents on the B and D rings.<sup>6</sup> Because of its unique tetracyclic structure, many groups are engaged in synthetic studies of the fawcettimine-type alkaloids.<sup>7,8</sup> The main focus of these studies has been the synthesis of the hydrindane skeleton (B and D rings), and they are based on Inubushi and Heathcock's pioneering work.<sup>8p,q</sup> As an exceptional strategy, Dake and Kozak reported Pt(II)-catalyzed cyclization to construct an octahydroquinoline skeleton having a quaternary carbon (C and D rings).<sup>8n</sup> While Dake's synthesis proposed the possibility of an octahydroquinoline skeleton as a useful intermediate. Additionally, the

skeleton has been found in various alkaloids such as fendleridine.<sup>9</sup> Therefore, the development of a concise synthetic method for these skeleta would contribute to the synthesis of not only fawcettimine-type alkaloids but also related natural products.

Considering a synthesis of various analogs, we envisioned that an intramolecular cyclopropanation of tetrahydropyridine **1** followed by a ring-opening would give both octa- and decahydroquinoline **4** and **5**, respectively (Scheme 1). That is, by introducing an electron-withdrawing group as a substituent ( $R^3$  = EWG), a ring-opening of cyclopropane **2** would readily produce an iminium intermediate **3**,<sup>10</sup> which would be converted to **4** and **5** by a proton transfer and stereoselective reduction, respectively. It was expected that various substituents could be introduced into the obtained compounds **4** and **5** by using ketone functionality for the synthesis of fawcettimine-type alkaloids. Moreover, the iminium intermediate **3**, derived from a substrate having a hydroxyethyl group ( $R^2$  =  $\text{CH}_2\text{CH}_2\text{OH}$ ), would be intramolecularly trapped with an alcohol to give a tricyclic compound **6**, which can be found in a fendleridine skeleton. While there are many reports about cyclopropanation of mono-substituted enamides and enecarbamates,<sup>11,12</sup> the number of reports decreases as the number of substituents increases. In the case of tetrasubstituted enamides and enecarbamates, pyrolysis of tosylhydrazone<sup>13a</sup> and formation of carbene from chloroform via treatment of a strong base<sup>13b</sup> were only employed for cyclopropanation.<sup>14</sup> Thus, there is no example of the use of a transition metal-carbene complex. Herein, we report synthesis of octahydroquinolines **4**, decahydroquinolines **5**, and tricyclic compound **6** having a quaternary carbon center at the angular position via a transition metal-catalyzed cyclopropanation of tetrasubstituted enecarbamate and a ring-opening.



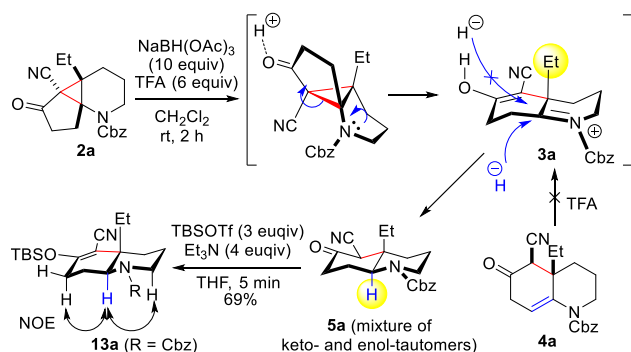
**Figure 1.** Fawcettimine-type alkaloids and a natural product containing both octa- and decahydroquinoline skeletons.

entry	Lewis acid or H <sup>+</sup>	reductant	time	yield (%) <sup>a</sup>	
				<b>4a</b>	<b>5a</b>
1	TFA <sup>b</sup>	none	30 min	96	0
2	BF <sub>3</sub> ·OEt <sub>2</sub> <sup>b</sup>	none	30 min	95	0
3	TFA	Et <sub>3</sub> SiH	2 h	56	0

4	TFA	Ph <sub>3</sub> SiH	5 h	40	0
5	TFA	NaBH <sub>3</sub> CN	2 h	17	0
6	TFA	NaBH(OAc) <sub>3</sub>	2 h	22 <sup>c</sup>	72 <sup>c</sup>
7	BF <sub>3</sub> ·OEt <sub>2</sub> <sup>d</sup>	NaBH(OAc) <sub>3</sub> <sup>e</sup>	11 h	23 <sup>c</sup>	31 <sup>c</sup>
8	Ti(OiPr) <sub>4</sub>	NaBH(OAc) <sub>3</sub>	15 min	29 <sup>f</sup>	0
9	AlCl <sub>3</sub>	NaBH(OAc) <sub>3</sub>	3 h	59	0
10	Sc(OTf) <sub>3</sub>	NaBH(OAc) <sub>3</sub>	2 h	28	0
11	MgBr <sub>2</sub> ·OEt <sub>2</sub>	NaBH(OAc) <sub>3</sub>	3 h	60	0

a. Isolated yield. b. 1 equiv. c. The yield was calculated by <sup>1</sup>H NMR. d. 6 equiv. e. 10 equiv. f. The starting material was recovered (31%). TFA = trifluoroacetic acid.

**Scheme 3. Stereoselectivity of the reduction of iminium intermediate 3a.**



To determine the scope and limitations, the optimized conditions were applied for the synthesis of several octahydroquinolines **4b–4h** (Table 3). Cyclopropanation of substrates **1b** and **1c** having CO<sub>2</sub>Me and Boc group as protecting groups gave compounds **2b** and **2c** in moderate yields, respectively (entries 1 and 2). Compound **2b** readily converted to octahydroquinoline **4b** when treated with BF<sub>3</sub>·OEt<sub>2</sub>. In the case of **2c**, MgBr<sub>2</sub>·OEt<sub>2</sub> was used instead of BF<sub>3</sub>·OEt<sub>2</sub> for suppressing the undesired deprotection of the Boc group. Compound **4c** was obtained in excellent yield. Interestingly, the reaction of **1d** with its tosyl group gave **4d** in 76% yield without acidic treatment, as the corresponding cyclopropane **2d** was easily cleaved (entry 3). The reaction of β-keto-α-diazoester **1e** did not give compound **2e**. Thus a nitrile group is essential for this cyclopropanation (entry 4). The reaction of **1f** did not give cyclopropane **2f** because C–H insertion was competitive (entry 5). Benzyl (Bn) and TBS groups were compatible under these reactions, and octahydroquinolines **4g** and **4h** were obtained in 80% and quantitative yields, respectively (entries 6 and 7).

**Table 3. Scope and limitations of synthesis of octahydroquinoline 4.**

entry	first step	yield <sup>a</sup>	second step	yield <sup>a</sup>
1		<b>2b:</b> 60%		<b>4b:</b> 96%

2		<b>2c:</b> 50%		<b>4c:</b> quant <sup>b</sup>
3		<b>2d:</b> (-)		<b>4d:</b> 76% <sup>c,d</sup>
4		<b>2e:</b> 0%		<b>4e:</b> (-)
5		<b>2f:</b> 0%		<b>4f:</b> (-)
6		<b>2g:</b> 49%		<b>4g:</b> 80%
7		<b>2h:</b> 52%		<b>4h:</b> quant <sup>b</sup>

a. Isolated yield. b. MgBr<sub>2</sub>·OEt<sub>2</sub> was used instead of BF<sub>3</sub>·OEt<sub>2</sub>. c. Without acidic treatment. d. The reaction time was 35 min. Bn = benzyl, TBS = *tert*-butyldimethylsilyl.

Because compound **2** was not stable as described above, a one-pot cyclopropanation/reductive ring-opening was attempted for the synthesis of decahydroquinolines **5**. Treatment of **1a** with Rh<sub>2</sub>(esp)<sub>2</sub> was followed by reductive ring-opening using TFA and NaBH(OAc)<sub>3</sub> to give decahydroquinoline **5a** in 48% yield, which was comparable to the yield of the stepwise procedure (Table 4, entry 1). Methoxycarbonyl and Boc groups were compatible under these conditions (entries 2 and 3). In the case of substrates having Bn and TBS groups, the desired decahydroquinolines **5g** and **5h** were obtained as major products, respectively, while the reductive ring-opening competed with formation of octahydroquinolines **4g** and **4h**.

**Table 4. One-pot cyclopropanation and reductive ring-opening for synthesis of decahydroquinoline 5.**

entry	R <sup>1</sup>	R <sup>2</sup>	time	product	yield (%) <sup>a</sup>	ratio <sup>b</sup>
1 <sup>c</sup>	Cbz	Et	2 h	<b>5a</b>	48	4.0:1
2 <sup>d</sup>	CO <sub>2</sub> Me	Et	50 min	<b>5b</b>	64	7.3:1
3 <sup>d</sup>	Boc	Et	45 min	<b>5c</b>	77	6.5:1
4	Cbz	CH <sub>2</sub> CH <sub>2</sub> OBn	8 h	<b>5g</b>	47	3.3:1
5 <sup>d</sup>	Cbz	CH <sub>2</sub> CH <sub>2</sub> OTBS	2 h	<b>5h</b>	43	3.8:1

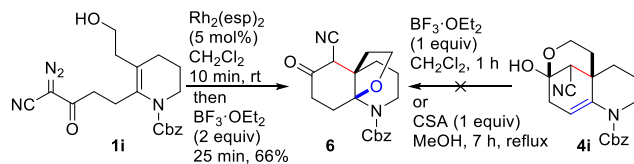
a. Isolated yield. b. The ratio was calculated by <sup>1</sup>H NMR. c. 0.1 mol% of Rh<sub>2</sub>(esp)<sub>2</sub> was used. d. Rh(OAc)<sub>2</sub> was used instead of Rh<sub>2</sub>(esp)<sub>2</sub>.

This one-pot procedure for decahydroquinolines **5** was extended to a one-pot synthesis of tricyclic compound **6** (Scheme



4). Treatment of  $\beta$ -keto- $\alpha$ -diazonitrile **1i** with  $\text{Rh}_2(\text{esp})_2$  followed by  $\text{BF}_3 \cdot \text{OEt}_2$  gave compound **6** via cyclopropanation in 66% yield. It is interesting that the cyclopropanation was the preferable process rather than O–H insertion in the first step, and an iminium ion could be trapped without formation of enecarbamate in the second step. It was difficult to obtain compound **6** from octahydroquinoline **4i** under mild conditions including treatment with  $\text{BF}_3 \cdot \text{OEt}_2$  or CSA.

#### Scheme 4. One-pot synthesis of tricyclic compound **6**.



In summary, we have investigated a concise synthesis of octahydroquinolines **4** and decahydroquinolines **5** having a quaternary carbon center at the angular position via a one-pot reaction involving cyclopropanation and ring-opening. The use of Ts-protected ene-sulfonamide **1d** was found to be effective for direct access to **4**. One-pot procedures for the synthesis of decahydroquinolines **5** and tricyclic compound **6** were also established. These methods would be powerful because a quaternary carbon center and a six-membered carbocycle were constructed at once. Synthesis of related alkaloids based on the developed strategy for decahydroquinolines **5** is now underway.

## ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website. Experimental procedures, spectra data ( $^1\text{H}$  and  $^{13}\text{C}$  NMR, IR, and HRMS) (PDF)

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### Notes

The authors declare no competing financial interest.

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